

*Figure 4* Bright-field TEM showing the base of scratches in (a) Mg-PSZ and (b) Ca-PSZ. Bar length equals five micrometer.

- PASCOE, *Nature* 253 (1975) 703.
- 2. R.T. PASCOE, R.H.J. HANNINK and R.C. GARVIE, Sci. Ceram. 9 (1977) 447.<br>GARVIE, Sci. Ceram. **9 (1977) 447.** *Received 1 September* **D** *L*. PORTER and A. H. HEUER *J. Amer. Ceram. and accepted 10 October 1980*
- 3. D. L. PORTER and A. H. HEUER, J. Amer. Ceram. *Soc.* 60 (1977) 183.
- *4. Iclem, ibid.* 62 (1979) 298.
- 5. S.T. GULATI, J.D. HELFINSTINE and A.D. B. T. GOLATI, **5. D. HELFINSTINE** and A. D. (2009) 211. R. H. J. HANNINK (2009) 211. R. H. J. HANNINK
- 6. R. C. GARVIE, R. H. J. HANNINK and C. URBANI, *Ceramurgica* 6 (1980) 19. *CSIRO Division of Materials Science*
- 7. R. HILL, "The Mathematical Theory of Plasticity" *Melbourne*  (Clarendon Press, Oxford, 1950). *Australia*
- References 8. R. H, J. HANNINK, J.. *Mater. Sci.* 13 (1978) 2407.
- 1. R. C. GARVIE, R. H. J. HANNINK and R. T. 9. M. L. MECARTNEY, W. T. DONLON and A. H. PASCOF Nature 253 (1975) 703.

## *Optical properties of palmitic acid thin films*

Long-chain fatty acid films find numerous applications in optical devices [1-5]. The optical properties of thin films of palmitic acid formed by the Langrnuir and Blodgett technique [6-9] and by vacuum evaporation [10] have been studied by various workers. In the present investigation the ion-plating technique [11] has been adopted to obtain the palmitic acid thin films and their optical properties are studied in detail.

Pure palmitic acid (99.5%; Eastmann, Kodak, New York) was evaporated from a molybdenum boat using an ion-plating technique 'in the presence of r.f. glow and deposited on to well-cleaned glass slides. The vacuum was maintained at about  $2 \times 10^{-2}$  Torr. The r.f. power was kept at 150W



*Figure 1* X-ray diffractograms for (a) bulk palmitic acid and (b) ion-plated palmitic acid film.

and the substrates were biased to 150V. The thickness of the films has been measured using a multiple-beam interferometer. A structural analysis of the film has been carried out using an X-ray diffractometer. A Shimadzu-UV-Vis doublebeam spectrophotometer was used to measure the transmittances of the films of various thicknesses (60-200 nm) in the visible region.

The absorption index  $(k_f)$  has been evaluated from Lambert's equation. The refractive index  $(n_f)$  has been determined from the following relation [ 12]

$$
T_0/4n_{\rm b} = \left[ (n_{\rm b} + 1) \cos \frac{2\pi d}{\lambda} (n_{\rm f} + ik_{\rm f}) - i \left( \frac{n_{\rm b}}{n_{\rm f} + ik_{\rm f}} + n_{\rm f} + ik_{\rm f} \right) \sin \frac{2\pi d}{\lambda} (n_{\rm f} + ik_{\rm f}) \right]^{-2}
$$

where  $T_0$  is the transmittance,  $k_f$  the absorption index,  $\lambda$  the wavelength of the light and  $n_b$  and  $n_f$ represent the refractive indices of the substrate and film respectively. The corresponding moduli have been used for the complex terms.

Palmitic acid films formed at room temperature were found to be transparent. Fig. 1 shows the X-ray diffractrograms for (a) palmitic acid powder and (b) the ion-plated film. From the figure it can be seen that the crystalline structure is maintained without any decomposition due to the evaporation process. From Fig. 2 the transmittance in the visible region has been found to be fairly high  $(> 90%)$  and increases slowly with wavelength. At higher wavelengths it tends to become constant thereby suggesting that the films are less absorbing at higher wavelength. The variation of log (transmittance) with film thickness is represented in Fig. 3. From the plot, the absorption index  $(k_f)$ and absorption coefficient  $(\alpha)$  have been estimated. The absorption indices in the region 700-400nm (Fig. 4) have been found to be extremely low (the value lies between 0.0095 and 0.0145) and hence have been neglected in the



*Figure 2* Variation of transmittance with wavelength for films of different thicknesses.

evaluation of  $n_f$ . Fig. 4 shows the variation of  $n_f$ with  $\lambda$  for films of different thicknesses: it is found that thinner films have lower values of refractive indices. This may be due to the presence of voids and discontinuities in thinner films [ 13 ].

It is interesting to note that the refractive index (1.551) of the film (62.5 nm) at a wavelength of 600nm is quite comparable with that (1.557) obtained for monolayer film of thickness 2.2 nm [14, 15]. The higher values of the refractive indices



*Figure 3* Variation of log T with film thickness d.

of the films in comparison with that of the bulk material [16] (1.4335) may be due to the porous nature of the film [17-20] and rise of substrate temperature [21-23] during the film formation. It is well known that the highly transparent films will have their absorption edge in the u.v. region [24]. This is in conformity with the present observation (Fig. 2) made with palmitic acid films. From the  $\alpha^2$  versus *hv* plot (Fig. 5) the value



*Figure 4* Variation of  $n_f$  and  $k_f$  with wavelength  $\lambda$ .



*Figure 5* Variation of  $\alpha^2$  with photon energy hv.

of  $E_{\text{opt}}$  has been estimated as 2.2 eV [25, 26]. As the glass does not show any absorption in the u.v. and visible regions above 320nm (Fig. 2), it can be inferred that the value of  $E_{opt}$  is not affected by the glass substrate.

## **Acknowledgements**

One of the authors (ARR) is thankful to Sri Ramakrishna Mission Vidyalaya Arts College, Coimbatore and the University Grants Commission, New Delhi, for assistance under the Faculty Improvement Programme.

## **References**

- *1. K. B. BLODGETT, Phys. Rev.* 55 (1939) 391.
- 2. I. LANGMUIR, *Proc. Roy. Soc., London* Series A. 170 (1939) 1.
- 3. R. ULRICH and H.P. WEBER, *AppL Opt.* **11**  (1972) 428.
- 4. V.K. AGARWAL, *Thin Solid Films 50* (1978) 3.
- 5. T.P. SOSNOWSKI and H. P. WEBER, *Opt. Commun*  7 (1973) 47.
- 6. V.K. SRIVASTAVA, in "Physics of Thin Films", Vol. 7 (edited by G. Hass, M. H. Francombe and R. W. Hoffman) (Academic Press, New York, 1973) p. 311.
- 7. D.D. ENGELSEN, Jr. *Opt. Soc. Amer.* 61 (1971) 1460.
- 8. D.D. ENGELSEN and B. DEKONIG, *J. Chem. Soc. Faraday 1* 70 (1974) 1603.
- 9. M.S. TOMAR and V.K. SRIVASTAVA, *Ind. J. Pure Appl. Phys.* 10 (1972) 573.
- 10. A.R. KASILINGAM, PhD thesis, University of Madras, India (1979).
- 11. Y. MURAYAMA and T. TAKAO, *Thin Solid Films*  40 (1977) 309.
- 12. G. HASS and C. D. SALZBERG, J. Opt. Soc. Amer. 44 (1954) 181.
- 13. A. GOSWAMI and A.P. GOSWAMI, *Thin Solid Films* 16 (1973) 175.
- 14. D.D. ENGELSEN, *Surf. Sci.* 56 (1976) 272.
- 15. G. T. AYOUB and N. M. BASHARA, J. Opt. Soc. *Amer.* 68 (1978) 978.
- 16. R.C. WEAST (editor) "Handbook of Chemistry and Physics", 57th edition, (CRC Press, Cleveland, Ohio, 1976-77) C-331.
- 17. R.M. HANDY and L.C. SCALA, J. *Electrochem. Soc.* 113 (1966) 109.
- 18. H. *KOCH, Phys. Stat. Solidi* 12 (1965) 533.
- 19. E. RITTER and R. HOFFMANN, J. Vac. Sci. *Technol.* 6 (1969) 733.
- 20. S. OGURA, N. SUGAWARA and R. HIRAGA, *Thin Solid Films* 30 (1975) 3.
- 21. D.M. MATTOX, J. *Electrochem. Technol.* 2 (1964) 294.
- 22. G. HASS, M.H. FRANCOMBE and R.W. HOFF-MANN (editors) "Physics of Thin Films", Vol. 8 (1975) 6.
- 23. G. HASS and E. RITTER, J. Vac. Sci. Technol. 4 (1967) 71.
- 24. A. GOSWAMI and A.P. GOSWAMI, *Thin Solid Films* 27 (1975) 123.
- 25. M. BALOG and M. SCHIEBA, M. MICHMANN and S. PATAI, *Thin SolidFilms* 41 (1977) 247.
- 26. H.Y. *FAN, Rep. Prog. Phys.* 19 (1956) 107.

*Received 5 September and accepted 13 October 1980* 

> A. R. RAMARAJU C. BALASUBRAMANIAN M. RADHAKRISHNAN *Department of Physics, Madras University Autonomous P. G. Centre, Coimbatore - 641041, India*